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Important notes: • Hydrogen and the metals form positive ions • Non-metals form negative ions, and their names end in • Group 4 and 5 do not usually form ions because they would have to lose or gain several electrons and that takes too much energy -ide • Group 0 elements do ...

IGCSE - Chemistry

IGCSE Complete Chemistry Notes According to Syllabus IGCSE Chemistry Syllabus Compilation by : WooWooWoo 2 Compiled by WooWooWoo
Chapter 1: The Particulate nature of matter Matter is anything that has mass and occupies space There are 3 states of matter, solids, liquids and gases

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Pure substances and impurities A pure substance is a substance that has no particles of any other substance mixed with it An unwanted substance, mixed with a wanted substance, is called an impurity To check if a substance is pure, you have to check its melting and boiling points

Interactive Learner Guide - CIE Notes

14 Organic chemistry Names and properties of organic compounds In addition to the syllabus content, you are also expected to understand and know experimental skills For Papers 1-4 and Paper 6, you will also need to learn a number of tests and test results for different ions and gases called the 'Notes

Complete Chemistry www.aswarphysics.weebly

Complete Chemistry for Cambridge IGCSE If you are taking IGCSE chemistry, using the Cambridge International Examinations syllabus 0620, then this book is for you Look at these notes: Core curriculum If you are following the Core curriculum, you can ignore any material with a red line beside it

TEACHER'S NOTES I.G.C.S.E. CHEMISTRY

CHAPTER - ACIDS, BASES AND SALTS IGCSE CHEMISTRY PAGE|8-1 T Chapter 8 Acids, Bases and Salts

IGCSE Chemistry 2012 exam revision notes

IGCSE Chemistry 2012 exam revision notes by Samuel Lees Contents: 1 The particulate nature of matter 2 Experimental techniques 21 Measurement 22 (a) Criteria of purity 22 (b) Methods of purification 3 Atoms, elements and compounds 31 Atomic structure and the Periodic Table 32 Bonding the structure of matter 32 (a) Ions and ionic bonds

PHYSICS IGCSE 2012 EXAM REVISION NOTES

PHYSICS IGCSE 2012 EXAM REVISION NOTES By Samuel Lees and Adrian Guillot 1 General physics 11 length and time 12 Speed, velocity and acceleration 13 Mass and weight 14 Density 15 Forces a Effects of forces b Turning effect c Conditions for equilibrium d Centre of mass e Scalars and vectors 16 Energy work power a Energy b

O-Level Sc (Chemistry) v1 - Yaaka Digital Network

Personal Tips: Chemistry is actually very easy! I used to get D7 for my Chemistry, but with perseverance for 3 months, I scored A1 in my O-Levels 2008! - If you can't do a question, use a red-pen to circle it, fold the page! Arrive school earlier, clear doubts with teachers outside the staffroom

Mathematics IGCSE notes Index - WELCOME IGCSE

1 Decimals and standard form top (a) multiplying and dividing (i) Move the decimal points to the right until each is a whole number, noting the total number of moves, perform the multiplication, then

ORGANIC CHEMISTRY - uCoz

www.igcseatua www.igcseatua ORGANIC CHEMISTRY OIL and its many useful PRODUCTS The origin of oil Crude oil is formed from organic material of the remains of plant and animal organisms that lived millions of years ago These remains form sediments eg at the bottom of seas, and become buried under layers of sedimentary rock

TABLE OF CONTENTS - Peterhouse Boys' School

June 2014 CIE IGCSE GEOGRAPHY (0460) Zubair Junjunia Page 5 of 19 Threshold Population: The minimum amount of people required for a service to be offered and remain open High Order Goods (Comparison): Goods that people buy less frequently They tend to be more expensive and people will normally compare quality and price before purchasing eg a car

BASIC CHEMISTRY

Basic Chemistry 1/1/2018©TLC (866) 557-1746 9 CEU COURSE DESCRIPTION BASIC CHEMISTRY CEU TRAINING COURSE This five hour continuing education course is a non-lab course that introduces the students to basic chemistry principles Topics covered include the periodic table,

atoms and molecules, the pH scale and experimental design and techniques

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